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(54) **Thermal barrier coating material comprising rare earth oxides**

(57) A coating material for a component intended for use in a hostile thermal environment has a cubic microstructure and consists essentially of either zirconia stabilized by dysprosia, erbia, gadolinium oxide, neodymia, samarium oxide or ytterbia, or hafnia stabilized by dys-

prosia, gadolinium oxide, samarium oxide, yttria or ytterbia. Up to five weight percent yttria may be added to the coating material.

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Description

[0001] This invention generally relates to coatings for components exposed to high temperatures, such as the hostile thermal environment of a gas turbine engine. More particularly, this invention is directed to a protective coating for a thermal barrier coating (TBC) on a gas turbine engine component, in which the protective coating has a low thermal conductivity, and may be resistant to infiltration by contaminants present in the operating environment of a gas turbine engine.

[0002] Higher operating temperatures for gas turbine engines are continuously sought in order to increase their efficiency. However, as operating temperatures increase, the high temperature durability of the components within the hot gas path of the engine must correspondingly increase. Significant advances in high temperature capabilities have been achieved through the formulation of nickel and cobalt-base superalloys. Nonetheless, certain components of the turbine, combustor and augmentor sections of a gas turbine engine can be required to operate at temperatures at which the mechanical properties of such alloys are insufficient. For this reason, these components are often protected by a thermal barrier coating (TBC).

[0003] TBC's are typically formed of ceramic materials deposited by plasma spraying, flame spraying and physical vapor deposition (PVD) techniques. TBC's employed in the highest temperature regions of gas turbine engines are most often deposited by PVD, particularly electron-beam PVD (EBPVD), which yields a strain-tolerant columnar grain structure that is able to expand and contract without causing damaging stresses that lead to spallation. Similar columnar microstructures can be produced using other atomic and molecular vapor processes, such as sputtering (e.g., high and low pressure, standard or collimated plume), ion plasma deposition, and all forms of melting and evaporation deposition processes (e.g., cathodic arc, laser melting, etc.). In contrast, plasma spraying techniques such as air plasma spraying (APS) deposit TBC material in the form of molten "splats," resulting in a TBC characterized by a degree of inhomogeneity and porosity.

[0004] Various ceramic materials have been proposed as TBC's, the most notable of which is zirconia (ZrO_2) that is partially or fully stabilized by yttria (Y_2O_3), magnesia (MgO) or another alkaline-earth metal oxides, or ceria (CeO_2) or another rare-earth metal oxides to yield a tetragonal microstructure that resists phase changes. Still other stabilizers have been proposed for zirconia, including hafnia (HfO_2) (U.S. Patent No. 5,643,474 to Sangeeta) and gadolinia (gadolinium oxide; Gd_2O_3) (U.S. Patent No. 6,177,200 to Maloney). U.S. Patent Nos. 5,512,382 and 5,624,721 to Strangman mention yttria-stabilized hafnia as a possible TBC material, though neither of these patents suggests what a suitable composition or microstructure might be. Still other proposed TBC materials include ceramic materials with the pyrochlore structure $\text{A}_2\text{B}_2\text{O}_7$, where A is lanthanum, gadolinium or yttrium and B is zirconium, hafnium and titanium (U.S. Patent No. 6,117,560 to Maloney). However, yttria-stabilized zirconia (YSZ) has been the most widely used TBC material. Reasons for this preference for YSZ are believed to include its high temperature capability, low thermal conductivity, and relative ease of deposition by plasma spraying, flame spraying and PVD techniques.

[0005] To protect a gas turbine engine component from its hostile thermal environment, the thermal conductivity of a TBC is of considerable importance. Lower thermal conductivities enable the use of a thinner coating, reducing the weight of the component, and/or reduce the amount of cooling airflow required for air-cooled components such as turbine blades. Though the thermal conductivity of YSZ decreases with increasing yttria content, the conventional practice has been to partially stabilize zirconia with six to eight weight percent yttria (6-8%YSZ) to promote spallation resistance. Ternary YSZ systems have been proposed to reduce the thermal conductivity of YSZ. For example, commonly-assigned U.S. Patent Application Serial No. [Attorney Docket No. 13DV-13490] to Rigney et al. discloses a TBC of YSZ and alloyed to contain certain amounts of one or more alkaline-earth metal oxides (magnesia, calcia (CaO), strontia (SrO) and barium oxide (BaO)), rare-earth metal oxides (ceria, gadolinium oxide, lanthana (La_2O_3), neodymia (Nd_2O_3), and dysprosia (Dy_2O_3)), and/or such metal oxides as nickel oxide (NiO), ferric oxide (Fe_2O_3), cobaltous oxide (CoO), and scandium oxide (Sc_2O_3). According to Rigney et al., when present in sufficient amounts these oxides are able to significantly reduce the thermal conductivity of YSZ by increasing crystallographic defects and/or lattice strains. Another proposed ternary system based on YSZ and said to reduce thermal conductivity is taught in U.S. Patent No. 6,025,078 to Rickerby et al. The additive oxide is gadolinium oxide, dysprosia, erbia (Er_2O_3), europia (Eu_2O_3), praseodymia (Pr_2O_3), urania (UO_2) or ytterbia (Yb_2O_3), in an amount of at least five weight percent to reduce phonon thermal conductivity.

[0006] Additions of oxides to YSZ coating systems have also been proposed for purposes other than lower thermal conductivity. For example, U.S. Patent No. 4,774,150 to Amano et al. discloses that bismuth oxide (Bi_2O_3), titania (TiO_2), terbia (Tb_4O_7), europia and/or samarium oxide (Sm_2O_3) may be added to certain layers of a YSZ TBC for the purpose of serving as "luminous activators."

[0007] To be effective, a TBC must strongly adhere to the component and remain adherent throughout many heating and cooling cycles. The latter requirement is particularly demanding due to the different coefficients of thermal expansion (CTE) between ceramic materials and the substrates they protect, which as noted above are typically superalloys, though ceramic matrix composite (CMC) materials are also used. An oxidation-resistant bond coat is often employed

to promote adhesion and extend the service life of a TBC, as well as protect the underlying substrate from damage by oxidation and hot corrosion attack. Bond coats used on superalloy substrates are typically in the form of an overlay coating such as MCrAlX (where M is iron, cobalt and/or nickel, and X is yttrium or another rare earth element), or a diffusion aluminide coating. During the deposition of the ceramic TBC and subsequent exposures to high temperatures, such as during engine operation, these bond coats form a tightly adherent alumina (Al_2O_3) layer or scale that adheres the TBC to the bond coat.

[0008] The service life of a TBC system is typically limited by a spallation event brought on by thermal fatigue. In addition to the CTE mismatch between a ceramic TBC and a metallic substrate, spallation can be promoted as a result of the TBC being contaminated with compounds found within a gas turbine engine during its operation. A notable example is a mixture of several different compounds, typically calcia, magnesia, alumina and silica, referred to herein as CMAS. CMAS has a relatively low melting eutectic (about 1190°C) that when molten is able to infiltrate to the cooler subsurface regions of a TBC, where it resolidifies. During thermal cycling, the CTE mismatch between CMAS and the TBC promotes spallation, particularly TBC deposited by PVD and APS due to the ability of the molten CMAS to penetrate their columnar and porous grain structures, respectively.

[0009] It would be desirable if improved TBC materials were available that exhibited low thermal conductivities, and preferably also exhibited resistance to spallation attributable to CMAS infiltration.

[0010] The present invention generally provides a coating material, particularly a thermal barrier coating (TBC), for a component intended for use in a hostile thermal environment, such as the superalloy turbine, combustor and augmentor components of a gas turbine engine. The coating material has a cubic microstructure and consists essentially of either zirconia (ZrO_2) stabilized by dysprosia (Dy_2O_3), gadolinium oxide (Gd_2O_3), erbia (Er_2O_3), neodymia (Nd_2O_3), samarium oxide (Sm_2O_3) or ytterbia (Yb_2O_3), or hafnia (HfO_2) stabilized by dysprosia, gadolinium oxide, samarium oxide or ytterbia. Up to five weight percent yttria may be added to the coating materials to further promote thermal cycle fatigue life.

[0011] According to the invention, zirconia and hafnia alloyed with their respective above-noted stabilizers have been shown to have lower thermal conductivities than conventional 6-8%YSZ, allowing for the use of a thinner coating and/or lower cooling airflow for air-cooled components. In addition, the hafnia-based coatings of this invention are resistant to infiltration by CMAS, thereby promoting the life of the TBC by reducing the risk of CMAS-induced spallation. While others have proposed additions of some of the oxides used as stabilizers in the present invention, including the aforementioned U.S. Patent Application Serial No. 09/833,446 to Rigney et al., U.S. Patent No. 6,025,078 to Rickerby et al., U.S. Patent No. 6,117,560 to Maloney and U.S. Patent No. 4,774,150 to Amano et al., such prior uses were based on additional oxides present in limited regions of a TBC (Amano et al.), or oxides added to the binary YSZ system in which zirconia is stabilized by yttria to yield a tetragonal microstructure (Rigney et al. and Rickerby et al.) or a cubic pyrochlore microstructure (Maloney) which therefore differ from the cubic (fluorite-type) microstructures of the present invention.

[0012] The coatings of this invention can be readily deposited by PVD to have a strain-resistant columnar grain structure, which reduces the thermal conductivity and promotes the strain tolerance of the coating. Alternatively, the coatings can be deposited by plasma spraying to have microstructures characterized by splat-shaped grains.

[0013] The invention will now be described in greater detail, by way of example, with reference to the drawings, in which:-

Figure 1 is a perspective view of a high pressure turbine blade.

Figure 2 schematically represents a cross-sectional view of the blade of Figure 1 along line 2--2, and shows a thermal barrier coating system on the blade in accordance with a preferred embodiment of the invention.

[0014] The present invention is generally applicable to components subjected to high temperatures, and particularly to components such as the high and low pressure turbine nozzles and blades, shrouds, combustor liners and augmentor hardware of gas turbine engines. An example of a high pressure turbine blade 10 is shown in Figure 1. The blade 10 generally includes an airfoil 12 against which hot combustion gases are directed during operation of the gas turbine engine, and whose surface is therefore subjected to hot combustion gases as well as attack by oxidation, corrosion and erosion. The airfoil 12 is protected from its hostile operating environment by a thermal barrier coating (TBC) system schematically depicted in Figure 2. The airfoil 12 is anchored to a turbine disk (not shown) with a dovetail 14 formed on a root section 16 of the blade 10. Cooling passages 18 are present in the airfoil 12 through which bleed air is forced to transfer heat from the blade 10. While the advantages of this invention are particularly desirable for high pressure turbine blades of the type shown in Figure 1, the teachings of this invention are generally applicable to any component on which a thermal barrier coating may be used to protect the component from a high temperature environment.

[0015] The TBC system 20 is represented in Figure 2 as including a metallic bond coat 24 that overlies the surface of a substrate 22, the latter of which is typically a superalloy and the base material of the blade 10. As is typical with

TBC systems for components of gas turbine engines, the bond coat 24 is preferably an aluminum-rich composition, such as an overlay coating of an MCrAlX alloy or a diffusion coating such as a diffusion aluminide or a diffusion platinum aluminide of a type known in the art. Aluminum-rich bond coats of this type develop an aluminum oxide (alumina) scale 28, which is grown by oxidation of the bond coat 24. The alumina scale 28 chemically bonds a TBC 26, formed of a thermal-insulating material, to the bond coat 24 and substrate 22. The TBC 26 of Figure 2 is represented as having a strain-tolerant microstructure of columnar grains 30. As known in the art, such columnar microstructures can be achieved by depositing the TBC 26 using a physical vapor deposition technique, such as EBPVD. The invention is also believed to be applicable to noncolumnar TBC deposited by such methods as plasma spraying, including air plasma spraying (APS). A TBC of this type is in the form of molten "splats," resulting in a microstructure characterized by irregular flattened grains and a degree of inhomogeneity and porosity.

[0016] As with prior art TBC's, the TBC 26 of this invention is intended to be deposited to a thickness that is sufficient to provide the required thermal protection for the underlying substrate 22 and blade 10, generally on the order of about 75 to about 300 micrometers. According to the invention, the thermal-insulating material of the TBC 26 may be a two-component system of zirconia stabilized by dysprosia, gadolinium oxide, erbia, neodymia, samarium oxide or ytterbia, or a two-component system of hafnia stabilized by dysprosia, gadolinium oxide, samarium oxide, yttria or ytterbia. Three-component systems can be formed of these compositions by adding a limited amount of yttria, generally up to five weight percent, such as about 4 to about 5 weight percent. When formulated to have a cubic (fluorite-type) microstructure, each of these compositions has been shown by this invention to have a substantially lower thermal conductivity than YSZ, particular YSZ containing six to eight weight percent yttria. These compositions also have the advantage of having a relatively wide cubic region in their phase diagrams, such that impurities and inaccuracies in the coating chemistry are less likely to lead to a phase transformation. Based on an investigation discussed below, suitable, preferred and target chemistries (by atomic percent) for the TBC 26 are set forth below in Table I. These chemistries ensure a stable cubic microstructure over the expected temperature range to which the TBC 26 would be subjected if deposited on a gas turbine engine component.

TABLE I.

Base Material	Stabilizer	Stabilizer Content (at.%)	Stabilizer Content (at.%)
		Broad Range	Preferred Range
ZrO ₂	Dy ₂ O ₃	10 to 45%	10 to 30%
ZrO ₂	Er ₂ O ₃	10 to 25%	12 to 25%
ZrO ₂	Gd ₂ O ₃	10 to 25%	10 to 20%
ZrO ₂	Nd ₂ O ₃	8 to 22%	8 to 18%
ZrO ₂	Sm ₂ O	10 to 25%	10 to 20%
ZrO ₂	Yb ₂ O ₃	8 to 30%	15 to 25%
HfO ₂	Dy ₂ O ₃	10 to 50%	10 to 45%
HfO ₂	Gd ₂ O ₃	5 to 30%	10 to 25%
HfO ₂	Sm ₂ O	5 to 30%	10 to 20%
HfO ₂	Y ₂ O ₃	10 to 45%	15 to 40%
HfO ₂	Yb ₂ O ₃	10 to 50%	15 to 25%

[0017] In addition to low thermal conductivities, the hafnia-based compositions of Table I have also been shown to be resistant to the infiltration of CMAS. While not wishing to be held to any particular theory, it is believed that the high melting temperature and surface energy of hafnia leads to little or no bonding tendency to the CMAS eutectic composition, and therefore inhibits the infiltration and bonding of CMAS to the TBC 26 while CMAS is molten and therefore capable of infiltrating the TBC 26. To benefit from this capability, the hafnia-based coatings of this invention can be used alone or as the outermost layer of a multilayer TBC. Even when deposited by PVD to have a columnar grain structure as shown in Figure 2, the hafnia-based coating compositions of this invention have been observed to reject or minimize the formation and infiltration of CMAS that would otherwise result in a CTE mismatch that can lead to spallation of the TBC 26.

[0018] In an investigation leading to this invention, TBC's were deposited by EBPVD on specimens formed of the superalloy René N5 on which a PtAl diffusion bond coat had been deposited. The specimens were coated by evaporating a single ingot of the desired composition. The TBC's were deposited to have thicknesses on the order of about

75 to about 150 micrometers. The chemistries and thermal conductivities of the coatings are summarized in Table II below. Thermal conductivities are reported at about 890°C following both stabilization at about 1000°C and a thermal aging treatment in which the specimens were held at about 1200°C for about two hours to determine the thermal stability of their coatings.

TABLE II.

	Specimen	Stabilizer Content (at.%)	Stabilizer Content (wt.%)	Thermal Conductivity	
				Stabilized (W/mK)	Aged (W/mK)
	(Coating)	(at.%)	(wt.%)	(W/mK)	(W/mK)
	ZrO ₂ +Dy ₂ O ₃	15	34.8	1.13	1.19
	ZrO ₂ +Er ₂ O ₃	17	38.9	1.14	1.13
a	ZrO ₂ +Gd ₂ O ₃	19.6	41.0	0.95	1.21
b	ZrO ₂ +Gd ₂ O ₃	14.3	32.0	0.96	1.20
	ZrO ₂ +Nd ₂ O ₃	13	29.0	0.95	1.14
	ZrO ₂ +Sm ₂ O ₃	15	33.3	n/a	n/a
	ZrO ₂ +Yb ₂ O ₃	20	44.4	1.16	1.16
	ZrO ₂ +Yb ₂ O ₃	20	44.4	1.11	1.17
c	ZrO ₂ +Yb ₂ O ₃	19.5	43.0	0.95	1.03
d	ZrO ₂ +Yb ₂ O ₃	18.9	42.0	1.09	1.17
	HfO ₂ +Dy ₂ O ₃	30	43.2	0.84	0.96
	HfO ₂ +Gd ₂ O ₃	15	23.3	0.96	1.13
	HfO ₂ +Sm ₂ O ₃	20	29.3	n/a	n/a
	HfO ₂ +Y ₂ O ₃	30	31.5	n/a	n/a
	HfO ₂ +Yb ₂ O ₃	20	31.9	1.16	1.16

a - Further alloyed to contain 4 wt.% Y₂O₃ (about 3.1 at.%).

b - Further alloyed to contain 4.8 wt.% Y₂O₃ (about 3.4 at.%).

c - Further alloyed to contain 4 wt.% Y₂O₃ (about 3.2 at.%).

d - Further alloyed to contain 4.1 wt.% Y₂O₃ (about 3.2 at.%).

[0019] The above results evidenced that the zirconia and hafnia-based TBC coatings of this invention had much lower thermal conductivities than the industry standard 6-8%YSZ material (above about 1.6 W/mK), and are significantly more thermally stable than 7%YSZ in terms of the thermal conductivities. Based on these results, it is also believed that the thermal conductivities of the zirconia and hafnia-based compositions of this invention might be further reduced by the inclusion of third and/or fourth oxides. Suitable oxides for this purpose include those evaluated above, namely, dysprosia, gadolinium oxide, erbia, neodymia, samarium oxide and ytterbia, as well as potentially zirconia (for the hafnium-based compositions), hafnia (for the zirconia-based compositions), barium oxide (BaO), calcia (CaO), ceria (CeO₂), europia (Eu₂O₃), indium oxide (In₂O₃), lanthana (La₂O₃), magnesia (MgO), niobia (Nb₂O₅), praseodymia (Pr₂O₃), scandia (Sc₂O₃), strontia (SrO), tantalum (Ta₂O₅), titania (TiO₂) and thulia (Tm₂O₃).

[0020] For the sake of good order, various aspects of the invention are set out in the following clauses:-

1. A component (10) comprising an outer coating (26) having a cubic microstructure and consisting essentially of zirconia stabilized with dysprosia, erbia, neodymia, samarium oxide or ytterbia, or zirconia stabilized with gadolinium oxide and yttria, or hafnia stabilized with dysprosia, gadolinium oxide, samarium oxide or ytterbia.

2. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 45 atomic percent dysprosia.

3. A component (10) according to clause 1, wherein the outer layer (26) consists of zirconia stabilized by about 10 to about 30 atomic percent dysprosia.

4. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 25 atomic percent erbia.

5. A component (10) according to clause 1, wherein the outer layer (26) consists of zirconia stabilized by about 12 to about 25 atomic percent erbia.

6. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 25 atomic percent gadolinium oxide and up to about 5 weight percent yttria.

7. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 20 atomic percent gadolinium oxide and about 4 to about 5 weight percent yttria.

8. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 8 to about 22 atomic percent neodymia.

9. A component (10) according to clause 1, wherein the outer layer (26) consists of zirconia stabilized by about 8 to about 18 atomic percent neodymia.

10. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 25 atomic percent samarium oxide.

11. A component (10) according to clause 1, wherein the outer layer (26) consists of zirconia stabilized by about 10 to about 20 atomic percent samarium oxide.

12. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 8 to about 30 atomic percent ytterbia.

13. A component (10) according to clause 12, wherein the outer coating (26) further contains up to about 5 weight percent yttria.

14. A component (10) according to clause 1, wherein the outer coating (26) consists of zirconia stabilized by about 15 to about 25 atomic percent ytterbia and about 4 to about 5 weight percent yttria.

15. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of hafnia stabilized by about 10 to about 50 atomic percent dysprosia.

16. A component (10) according to clause 1, wherein the outer layer (26) consists of hafnia stabilized by about 10 to about 45 atomic percent dysprosia.

17. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of hafnia stabilized by about 5 to about 30 atomic percent gadolinium oxide.

18. A component (10) according to clause 1, wherein the outer layer (26) consists of hafnia stabilized by about 10 to about 25 atomic percent gadolinium oxide.

19. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of hafnia stabilized by about 5 to about 30 atomic percent samarium oxide.

20. A component (10) according to clause 1, wherein the outer layer (26) consists of hafnia stabilized by about 10 to about 20 atomic percent samarium oxide.

21. A component (10) according to clause 1, wherein the outer coating (26) consists of hafnia stabilized by about 10 to about 45 atomic percent yttria.

22. A component (10) according to clause 1, wherein the outer coating (26) consists of hafnia stabilized by about 15 to about 40 atomic percent yttria.

23. A component (10) according to clause 1, wherein the outer coating (26) consists essentially of hafnia stabilized

by about 10 to about 50 atomic percent ytterbia.

24. A component (10) according to clause 1, wherein the outer layer (26) consists of hafnia stabilized by about 15 to about 25 atomic percent ytterbia.

25. A component (10) according to clause 1, wherein the outer coating (26) contains about 4 to about 5 weight percent yttria.

26. A component (10) according to clause 1, further comprising a metallic bond coat (24) adhering the outer coating (26) to the component (10).

27. A component (10) according to clause 1, wherein the component (10) is a superalloy airfoil component (10) of a gas turbine engine.

Claims

1. A component (10) comprising an outer coating (26) having a cubic microstructure and consisting essentially of zirconia stabilized with dysprosia, erbia, neodymia, samarium oxide or ytterbia, or zirconia stabilized with gadolinium oxide and yttria, or hafnia stabilized with dysprosia, gadolinium oxide, samarium oxide or ytterbia.

2. A component (10) according to claim 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 45 atomic percent dysprosia.

3. A component (10) according to claim 1, wherein the outer layer (26) consists of zirconia stabilized by about 10 to about 30 atomic percent dysprosia.

4. A component (10) according to claim 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 25 atomic percent erbia.

5. A component (10) according to claim 1, wherein the outer layer (26) consists of zirconia stabilized by about 12 to about 25 atomic percent erbia.

6. A component (10) according to claim 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 25 atomic percent gadolinium oxide and up to about 5 weight percent yttria.

7. A component (10) according to claim 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 20 atomic percent gadolinium oxide and about 4 to about 5 weight percent yttria.

8. A component (10) according to claim 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 8 to about 22 atomic percent neodymia.

9. A component (10) according to claim 1, wherein the outer layer (26) consists of zirconia stabilized by about 8 to about 18 atomic percent neodymia.

10. A component (10) according to claim 1, wherein the outer coating (26) consists essentially of zirconia stabilized by about 10 to about 25 atomic percent samarium oxide.

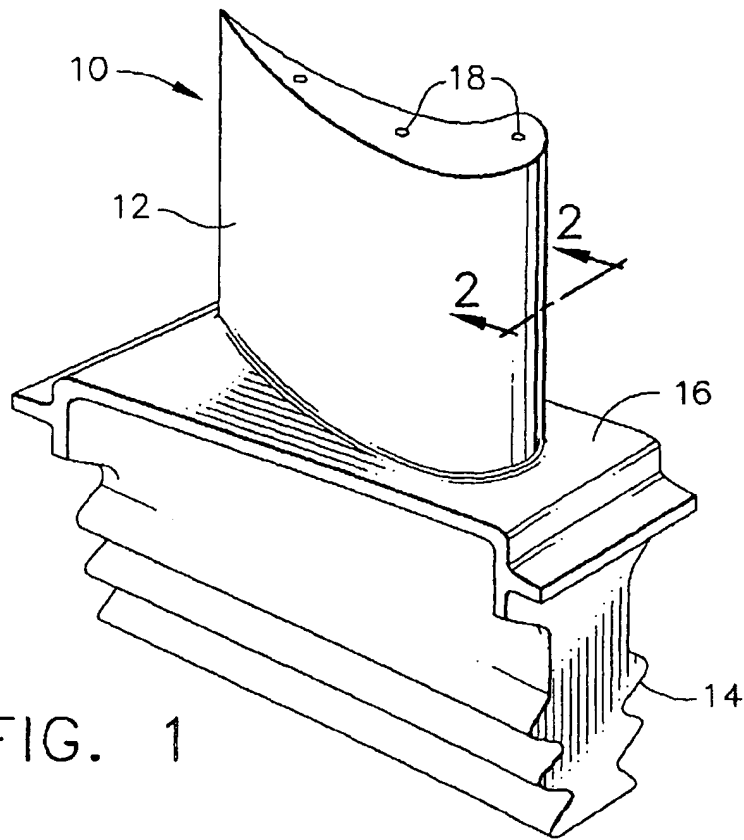


FIG. 1

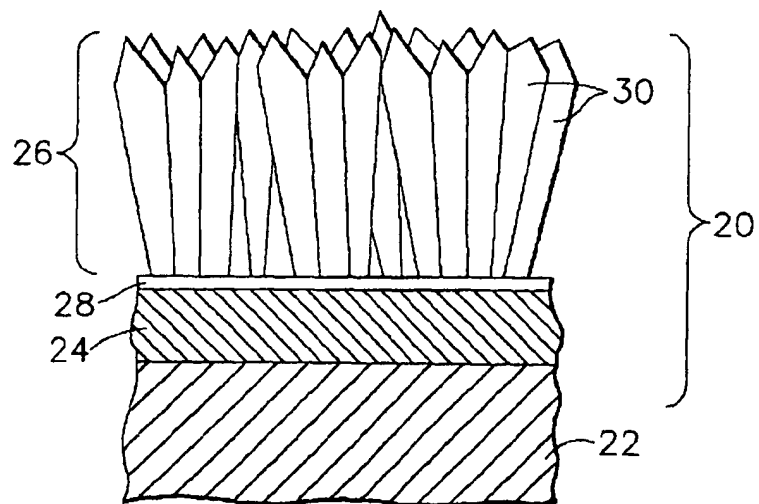


FIG. 2



European Patent
Office

EUROPEAN SEARCH REPORT

Application Number
EP 03 25 5387

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
Y,D	EP 0 848 077 A (UNITED TECHNOLOGIES CORP) 17 June 1998 (1998-06-17) * see p.2, lines 43/44, claim 1 * * the whole document *	1-10	C23C30/00 C23C16/00 C23C14/22 C23C14/08
X	EP 0 812 931 A (TOSOH CORP) 17 December 1997 (1997-12-17) * see p.3, l. 6-11, claim 1 * * the whole document *	1-10	
X	EP 0 595 451 A (PRAXAIR TECHNOLOGY INC) 4 May 1994 (1994-05-04) * see p.3, l. 1/2, claim 3 * * the whole document *	1-10	
X	WO 99 42630 A (GADOW RAINER ;SCHAEFER GUENTER WILHELM (DE)) 26 August 1999 (1999-08-26) * see p.6, last par., claims 1 and 14 * * the whole document *	1-10	
X,D	EP 0 825 271 A (ROLLS ROYCE PLC) 25 February 1998 (1998-02-25) * see p.4, col. 5, l.28-35, claims * * the whole document *	1-10	TECHNICAL FIELDS SEARCHED (Int.Cl.7) C23C
X,D	US 4 774 150 A (TAKAHASHI YOSHIKAZU ET AL) 27 September 1988 (1988-09-27) * see col. 2, l. 25-41, claims * * the whole document *	1-10	
X	US 6 333 090 B1 (KROEDER CLAUS-JUERGEN ET AL) 25 December 2001 (2001-12-25) * see col. 7, l.28-33, claims 1,4,5,10 * * the whole document *	1-10	
The present search report has been drawn up for all claims			
Place of search MUNICH		Date of completion of the search 23 January 2004	Examiner Stellmach, J
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document</p>			



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DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	WO 02 14580 A (SIEMENS WESTINGHOUSE POWER) 21 February 2002 (2002-02-21) * see claims * * the whole document *	1-10	
Y	WO 97 01436 A (GEN ELECTRIC) 16 January 1997 (1997-01-16) * see claims * * the whole document *	1-10	
P,X, D	EP 1 249 515 A (GEN ELECTRIC) 16 October 2002 (2002-10-16) * see p.3, 15-18, claim 2 * * the whole document *	1-10	
P,X	EP 1 295 965 A (GEN ELECTRIC) 26 March 2003 (2003-03-26) * see claims 1 and 9 * * the whole document *	1-10	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
The present search report has been drawn up for all claims			
Place of search MUNICH		Date of completion of the search 23 January 2004	Examiner Stellmach, J
<p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document</p>			

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

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This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

23-01-2004

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 0848077	A	17-06-1998	US 6117560 A	12-09-2000
			DE 69719701 D1	17-04-2003
			DE 69719701 T2	04-12-2003
			EP 0848077 A1	17-06-1998
			JP 10212108 A	11-08-1998
			US 2003049470 A1	13-03-2003
			US 6177200 B1	23-01-2001
			US 6231991 B1	15-05-2001
			US 6284323 B1	04-09-2001
			US 2001007719 A1	12-07-2001
EP 0812931	A	17-12-1997	DE 69700448 D1	30-09-1999
			DE 69700448 T2	13-01-2000
			EP 0812931 A1	17-12-1997
			JP 10081957 A	31-03-1998
			US 5789330 A	04-08-1998
EP 0595451	A	04-05-1994	US 5304519 A	19-04-1994
			CA 2094531 A1	29-04-1994
			CN 1086234 A ,B	04-05-1994
			DE 69302252 D1	23-05-1996
			DE 69302252 T2	19-09-1996
			EP 0595451 A1	04-05-1994
			JP 2824189 B2	11-11-1998
			JP 6135768 A	17-05-1994
			KR 136554 B1	25-04-1998
			US 5466208 A	14-11-1995
WO 9942630	A	26-08-1999	US 5418015 A	23-05-1995
			DE 19807163 C1	28-10-1999
			AT 218626 T	15-06-2002
			DE 59901628 D1	11-07-2002
			WO 9942630 A1	26-08-1999
			EP 1060281 A1	20-12-2000
			ES 2177251 T3	01-12-2002
			JP 2002504627 T	12-02-2002
			PT 1060281 T	31-10-2002
			US 6602814 B1	05-08-2003
EP 0825271	A	25-02-1998	DE 69713001 D1	11-07-2002
			DE 69713001 T2	27-02-2003
			EP 0825271 A1	25-02-1998
			US 6025078 A	15-02-2000
US 4774150	A	27-09-1988	JP 62207885 A	12-09-1987

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 03 25 5387

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.
The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

23-01-2004

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 6333090	B1	25-12-2001	NONE	
WO 0214580	A	21-02-2002	US 6258467 B1	10-07-2001
			CA 2417212 A1	21-02-2002
			EP 1309738 A2	14-05-2003
			WO 0214580 A2	21-02-2002
			US 2002061416 A1	23-05-2002
WO 9701436	A	16-01-1997	CH 690582 A5	31-10-2000
			DE 19680503 T0	08-01-1998
			IN 187269 A1	16-03-2002
			JP 10505299 T	26-05-1998
			WO 9701436 A1	16-01-1997
			US 5914189 A	22-06-1999
EP 1249515	A	16-10-2002	US 2002172838 A1	21-11-2002
			BR 0201230 A	31-12-2002
			EP 1249515 A2	16-10-2002
EP 1295965	A	26-03-2003	US 2003059633 A1	27-03-2003
			EP 1295965 A2	26-03-2003